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REACTION OF DICHLOROCARBENE WITH SOME ORGANO-PHOSPHORUS COMPOUNDS IN LOW COORDINATION STATES

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Abstract The reactions of dichlorocarbene with some organophosphorus compounds in low coordination states such as diphosphenes, phosphaethylenes, phosphaallenes, phosphabutatrienes, and phosphaalkynes were carried out.

INTRODUCTION

We have been successful in isolating organophosphorus compounds in low coordination states by introducing a bulky substituent such as 2,4,6-tri-t-butylphenyl group (hereafter abbreviated to Ar) as a protecting auxiliary.² Kinetically stabilized phosphorus compounds thus obtained include diphosphenes,2 phosphaethylenes,3 phosphaallenes,4 diphosphaallenes,5 phosphabutatrienes,6,7 and phosphaalkynes.^{8,9} Moreover, we have been interested in the structures and reactivities of multiple bonds involving phosphorus atom(s) of coordination numbers 2 and 1. We describe here our experimental results on the reactions of dichlorocarbene with several phosphorus compounds in low coordination states.

Reaction of Dichlorocarbene with Diphosphenes

The reaction of dichlorocarbene with diphosphenes^{10,11} gave 3,3dichlorodiphosphiranes which were converted to the corresponding 1,3-diphosphaallenes with methyllithium or t-butyllithium and a typical example for ArP=PAr is shown as follows.

Reaction with Phosphaethylenes

The reaction of dichlorocarbene either with (E)- or (Z)-2-phenylphosphaethylene gave trans-phosphirane 10 which could be converted to 1-phosphaellene with t-butyllithium at low temperature, while methyllithium gave an alkylated product, 3-phospha-1-butyne. On the other hand, 2,2-diphenylphosphaethylene afforded a phosphaindane derivative carrying 1-chloro-2,2-diphenylethenyl group in the reaction with dichlorocarbene.

Reaction with Phosphaallenes

The reaction of 1-phosphaallene with dichlorocarbene gave 2,2-dichloro-3-diphenylmethylenephosphirane. The phosphirane was successfully converted to 4,4-diphenyl-1-phosphabutatriene with n-butyllithium at low temperature while t-butyllithium gave 1-phospha-2-butynes having one t-butyl group at either phosphorus atom.

$$P = C = C \xrightarrow{Ph} \qquad CCl_2 \qquad Ar \xrightarrow{Cl} C = CPh_2$$

$$\frac{n - BuLi}{P} \qquad P = C = C = CPh_2$$

Reaction with Diphosphabutatriene

The reaction of 1-phosphabutatriene gave ethenylidenephosphirane ¹⁴ and the resulting phosphirane was transformed to a phospha-[3]-radialene, 2,3-dimethylenephosphirane, ¹⁵ on heating in

refluxing hexane, the structure of which was confirmed by the X-ray crystallographic analysis.

$$\begin{array}{c} \text{Ar} \\ \text{P=C=C=CPh}_2 \xrightarrow{:\text{CCl}_2} & \text{Ar} \\ \text{Cl} \\ \text{Cl} & \text{Cl} \\ \text{Cl} & \text{Cl} \\ \hline & & \text{Ar} \\ & & \text{P} \xrightarrow{\text{C}} \\ \text{Ce=CPh}_2 \\ \end{array}$$

Reaction with Diphosphaallenes

The reaction of 1,3-diphosphaallene gave 3-dichloromethylene-1,2-diphosphirane 13 instead of the expected phosphinidenephosphirane. Furthermore the diphosphirane thus obtained was successfully converted to 1,4-diphosphabutatriene 14 with lithium naphthalenide, while the reaction with t- or s-butyllithium gave the corresponding butylated 1,4-diphospha-2-butyne. The structure of the methylenediphosphirane was unambiguously confirmed by the X-ray analysis.

Reaction with Phosphaalkynes

The reaction of dichlorocarbene with phosphaalkyne of coordination number 1, $ArC\equiv P$, gave ethynyl chloride probably via phosphirene intermediates, followed by elimination of $(PCl)_n$. Some other chloro- or bromocarbenes also gave similar reaction products. 17

$$Ar-C \equiv P \xrightarrow{:CCl_2} \begin{bmatrix} Cl \\ Ar-C \xrightarrow{C} P \end{bmatrix} \rightarrow \begin{bmatrix} Cl \\ Ar-C \xrightarrow{C} PCl \end{bmatrix} \xrightarrow{-(PCl)_n} Ar-C \equiv C-Cl$$

CONCLUSION

Thus the reactions or dichlorocarbene with organophosphorus compounds in low coordination states gave various kinds of interesting compounds some of which were useful in preparation of expanded phosphacumulenes with alkyllithium reagents or an electron transfer reagent.

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